



MEP Naturalhy Presentation

Explosion Hazard for Natural Gas/Hydrogen Mixtures

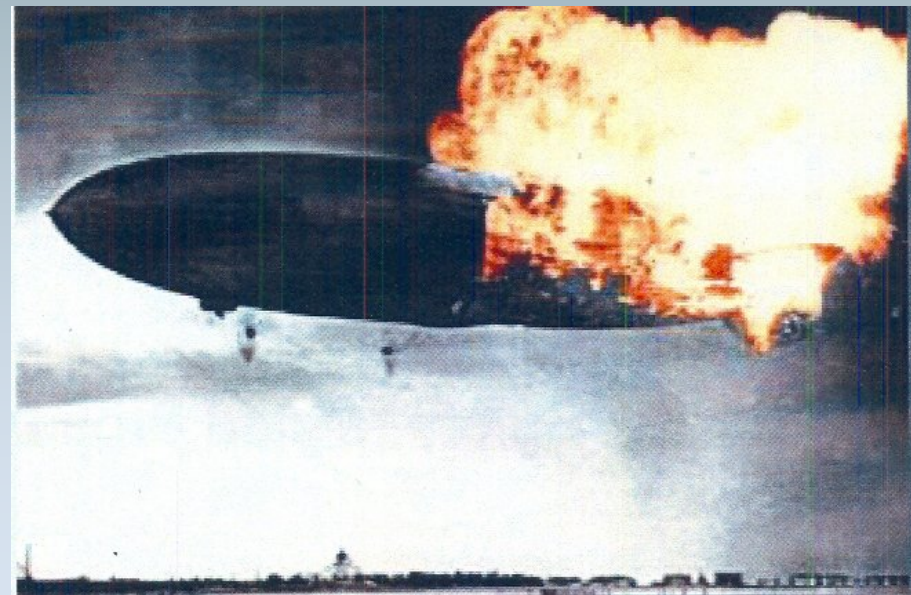
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The Explosion Problem

- Perception is that 'Hydrogen will explode'
- Historical events such as the Hindenburg disaster linger in the memory
- What about Natural Gas/
Hydrogen
Mixtures?





Naturalhy Explosions



- Confinement of walls creates pressure when gas ignited

- But without walls pressure can be generated by obstacles in path of flame-VCE

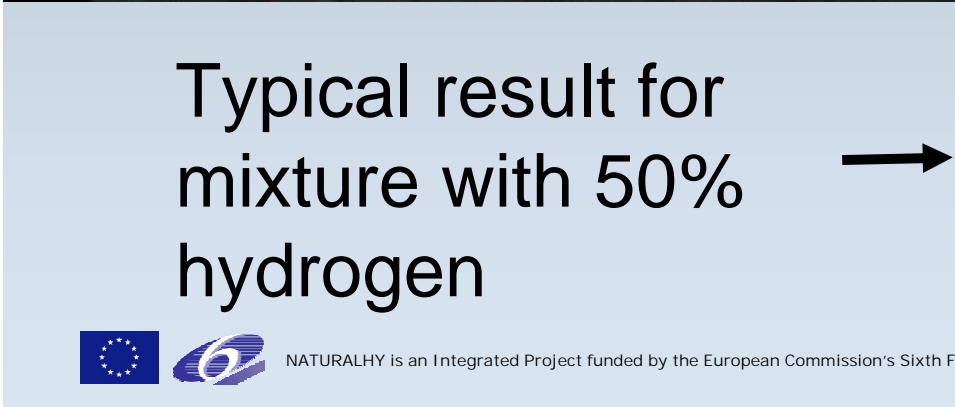




Explosions in Buildings



← Typical result for Natural Gas or Mixture with up to 20% Hydrogen



→ Typical result for mixture with 50% hydrogen





Vapour Cloud Explosions

- Presence of obstacles in flame path causes flame acceleration and high pressures
- Relevant to industrial sites
- Flixborough incident (1974) involving cyclohexane was classic example





Effect of Hydrogen Content

80:20



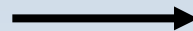
- Flame speed increases then decreases



50:50



- Continued flame acceleration





Conclusions

- Up to 30% H₂ can be added without significant increase in pressure compared to natural gas so unlikely to be an issue for Naturalhy concept
- For higher concentrations, an increase in explosion severity may arise
- Reducing obstacles will reduce explosion severity

